PRESSURE IN LIQUIDS, GASES, AND SOLIDS

Key Concepts

- Pressure is the force applied per unit area.
- · Pressure acts differently in liquids, gases, and solids.







Key Facts to Remember

- Pressure = Force ÷ Area, measured in pascals (Pa).
- In liquids, pressure increases with depth.
- Gases exert pressure when particles collide with container walls.
- Atmospheric pressure decreases with altitude.
- Solids apply pressure through contact surfaces, e.g., high heels exert more pressure than flat shoes.
- Hydraulic systems use liquid pressure to multiply force.
- Reducing area increases pressure (e.g., a knife blade).
- Vacuum packs remove air to reduce pressure inside the pack.

Quick Questions

- 1. What is the formula for pressure?
- 2. In which unit is pressure measured?
- 3. What happens to liquid pressure as depth increases?
- 4. What causes gas pressure?
- 5. Why does atmospheric pressure decrease with altitude?
- 6. Give an example of reducing area to increase pressure.
- 7. What do hydraulic systems use to multiply force?
- 8. Why are vacuum packs effective?

Fun Fact

At the bottom of the Mariana Trench, water pressure is over 1,000 times the pressure at sea level!

www.simplyscience.net